# Corrections

**Excess Gibbs Free Energies at Eight Temperatures and Excess Enthalpies and Volumes at $T$ = 298.15 K for Butanenitrile + 2-Butanol**. Rosa Garriga, Francisco Sánchez, Pascual Pérez, and Mariano Gracia, *J. Chem. Eng. Data* **1997**, *42*, 78−83.

Recently we found a mistake in $V^E$ measurements for {(1 − $x$)butanenitrile + $x$2-butanol}, which is important to correct in order to allow others to take advantage of our data.

The $V^E$ values that appear in the original Table 4 must be replaced by:

| $x$ | $V^E$/cm$^3$·mol$^{-1}$ | $x$ | $V^E$/cm$^3$·mol$^{-1}$ |
|---|---|---|---|
| 0.0285 | 0.046 | 0.5383 | 0.348 |
| 0.0778 | 0.122 | 0.7142 | 0.288 |
| 0.1850 | 0.239 | 0.7171 | 0.286 |
| 0.3208 | 0.318 | 0.8350 | 0.206 |
| 0.3633 | 0.334 | 0.8470 | 0.195 |
| 0.4757 | 0.351 | 0.9626 | 0.068 |
| 0.5205 | 0.349 | | |

$$V^E/\text{cm}^3\cdot\text{mol}^{-1} = x(1-x)\{1.396 + 0.078(1-2x) + 0.339(1-2x)^2\};\ \sigma(V^E) = 0.004$$

Consequently, Figure 3b must be modified, and the second to last sentence would be "$V^E$ is also lower for 1-butanol than for 2-butanol".

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**Vapor−Liquid Equilibria of Binary Mixtures Cyclopentane + Isopropyl Acetate, Isopropyl Acetate + Hexane, and Cyclopentane + Methyl Methacrylate at 101.3 kPa**. Kong-Wei Cheng, Jia-Yuh Chen, Muoi Tang, and Yan-Ping Chen, *J. Chem. Eng. Data* **1997**, *42*, 754−757.

The unit of the binary parameters $A_{12}$ and $A_{21}$ listed in Table 7 should be (cal mol$^{-1}$). It was misprinted as (J mol$^{-1}$) in the original article.

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**Phase Equilibria in the Systems Oxolane + Octane and Methyl 1,1-Dimethylethyl Ether + Hex-1-ene**. Jaime Wisniak, Eti Fishman, and Rotem Shaulovitch, *J. Chem. Eng. Data* **1998**, *43*, 304−306.

There is a misprint in the temperatures listed in Table 2. The correct values are

| $T$/K |
|---|
| 391.40 |
| 388.27 |
| 381.80 |
| 377.95 |
| 372.33 |
| 367.34 |
| 360.39 |
| 352.35 |
| 348.60 |
| 347.28 |
| 346.08 |
| 344.90 |
| 343.92 |
| 342.75 |
| 341.49 |
| 339.95 |
| 338.68 |
| 338.04 |

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